

1. Sulfonic acids
2. Organic derivatives of carbonic acid

Vladimíra Kvasnicová

Sulfonic acids

- functional group: $-\text{SO}_3\text{H}$
- names: *hydrocarbon sulfonic acid* / trivial names
- prefix: **sulfo-** (*the prefix used for $-\text{SH}$ group is sulfanyl- !!!*)
- crystalline or syrupy **water soluble** compounds
- **strong acids**
- formed by oxidation of thiols or by reaction of hydrocarbons with H_2SO_4
- their **sodium salts** are used as **detergents**

Sulfonic acids and their derivatives

- $\text{CH}_3\text{-SO}_3\text{H}$
- $\text{C}_6\text{H}_5\text{-SO}_3\text{H}$
- $\text{CH}_3\text{-(CH}_2)_2\text{-SO}_3\text{H}$
- $\text{CH}_3\text{-(CH}_2)_{15}\text{-SO}_3\text{H}$
- $\text{CH}_3\text{-CH(SO}_3\text{H)-CH}_2\text{-CH}_3$
- $\text{C}_6\text{H}_5\text{-SO}_3^- \text{Na}^+$
- methanesulfonic acid
- benzenesulfonic acid
- propanesulfonic acid
- hexadecane-1-sulf.a.
- butane-2-sulfonic ac.
- sodium benzenesulfonate

Sulfonic acids and their derivatives

- $\text{CH}_3\text{-SO}_2\text{Cl}$
- $\text{C}_6\text{H}_5\text{-SO}_2\text{-O-CH}_3$
- $\text{H}_2\text{N-CH}_2\text{-CH}_2\text{-SO}_3\text{H}$
- $\text{H}_2\text{N-C}_6\text{H}_4\text{-SO}_3\text{H}$
- $\text{R}_1\text{-HN-C}_6\text{H}_4\text{-SO}_2\text{NH-R}_2$
- methanesulfonyl chloride
- methyl benzenesulfonate
- 2-aminoethanesulfonic acid (= **taurine**)
- 4-aminobenzenesulfonic acid (= **sulfanilic acid**)

sulfonamides

Organic derivatives of carbonic acid

• carbonic acid: $\text{HO-CO-OH} = \text{H}_2\text{CO}_3$

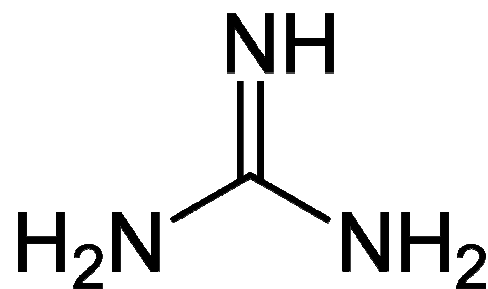
• carbamic acid: HO-CO-NH_2

• urea: $\text{H}_2\text{N-CO-NH}_2$

• thiourea: $\text{H}_2\text{N-CS-NH}_2$

• phosgene: Cl-CO-Cl *toxic gas*

• guanidine:



Exercise - repetition

- toluene
- m-cresol
- o-xylene
- naphthalene
- benzyl
- phenyl
- anthracene
- benzoic acid
- p-benzoquinone
- cyclohexanol
- acetaldehyde
- phenol
- acetone
- diethylether
- formaldehyde
- vinylchloride
- methanethiol
- hydroquinone